



MOCK CET – 3 PHYSICS, CHEMISTRY & MATHEMATICS

ANSWER KEYS

ANSWERS												
1	2	3	4	5	6	7	8	9	10	11	12	13
4	1	3	4	1	1	2	3	4	1	2	2	1
14	15	16	17	18	19	20	21	22	23	24	25	26
2	3	3	3	2	2	1	3	2	2	4	1	3
27	28	29	30	31	32	33	34	35	36	37	38	39
4	1	3	1	1	2	1	2	1	4	2	1	1
40	41	42	43	44	45	46	47	48	49	50	51	52
4	4	3	2	4	2	4	2	4	1	1	3	4
53	54	55	56	57	58	59	60					
4	2	1	1	2	3	1	1					

HINTS & SOLUTIONS

1. Here, $v = \frac{ds}{dt} = 3t^2 - 12t + 3$;

$$\Rightarrow a = \frac{dv}{dt} = 6t - 12$$

For $a = 0, 6t - 12 = 0$

$$\Rightarrow t = 2$$

At $t = 2s, v = 12 - 24 + 3 = -9ms^{-1}$

2. Form the equation of motion, $v^2 = u^2 - 2as$;

When $v = 0$, then the equation becomes

$$u^2 = 2as$$

$$\Rightarrow s \propto u^2$$

$$\Rightarrow \frac{S_1}{S_2} = \frac{u_1^2}{u_2^2}$$

$$\Rightarrow \frac{2}{S_2} = \frac{(40 \times 5/18)^2}{(80 \times 5/18)^2}$$

$$S_2 = 8 \text{ meters}$$

3. Here, $x = at^2$ and $v_x = \frac{dx}{dt} = 2at$;

$$v_y = \frac{dy}{dt} = \frac{d}{dt}(bt^2) = 2bt$$

$$\therefore v = \sqrt{v_x^2 + v_y^2} = \sqrt{4a^2t^2 + 4b^2t^2} = 2t\sqrt{a^2 + b^2}$$

4. Kinetic energy acquired by particle = $\frac{1}{2}mv^2$

Since particle travels distance d from rest then applying $v^2 = u^2 - 2as$

$$\Rightarrow v^2 = 2ad \text{ (since } u = 0)$$

$$\Rightarrow v^2 = 2 \frac{F}{m} d \text{ (since } F \text{ is constant)}$$

$$\text{Hence K.E} = \frac{1}{2}m \cdot \frac{2F}{m} d = Fd$$

5. According to Newton's first law of motion, the ball will revolve around the earth as no external force is applied.

6. According to conservation towards right, then

$$(m_1 + m_2)v = m_1u_1 + m_2u_2$$

$$\Rightarrow (m + 2m)v = m \times 3$$

$$\Rightarrow v = 1km/hour$$

7. Kinetic energy $= \frac{1}{2}mv^2$

Again, $v^2 = u^2 + 2as = 0 + 2ad$

Or $v^2 = 2\left(\frac{F}{m}\right)d$

$\therefore KE = \frac{1}{2}m \times 2\frac{F}{m}d = Fd$

$\therefore KE \propto d$

8. Given : $\vec{F} = (-2\hat{i} + 15\hat{j} + 6\hat{k})N$,

$\vec{d} = 10\hat{j}m$

Work done, $W = \vec{F} \cdot \vec{d}$

$= (-2\hat{i} + 15\hat{j} + 6\hat{k}) \cdot (10\hat{j})$

$= 150J$

9. Centripetal force, $F = mv_{max}^2$

$25 = \frac{0.25 \times v_{max}^2}{1.96}$

$\Rightarrow v_{max}^2 = \frac{25 \times 1.96}{0.25} = 196$

$\Rightarrow v_{max} = 14m/s$

10. Given : $\vec{F} = (-3\hat{i} + \hat{j} + 5\hat{k})$

$\vec{r} = (7\hat{i} + 3\hat{j} + \hat{k})$

Now, $\vec{\tau} = \vec{r} \times \vec{F}$

$\therefore \vec{\tau} = \begin{vmatrix} \hat{i} & \hat{j} & \hat{k} \\ 7 & 3 & 1 \\ -3 & 1 & 5 \end{vmatrix}$

$= \hat{i}(15 - 1) - \hat{j}(35 + 3) + \hat{k}(7 + 9)$

$= 14\hat{i} - 38\hat{j} + 16\hat{k}$

11. Escape velocity, $v_e = \sqrt{\frac{2GM}{R}} = 11.2km/sec$

And $v'_e = \sqrt{\frac{2GM}{R/4}} = 2\sqrt{\frac{2GM}{R}}$

$\Rightarrow v'_e = 2v_e$

$= 22.4km/sec$

12. Self explanatory

13. $T \propto \frac{1}{\sqrt{g}}$

Inside a satellite, $g = 0$

Hence period will be infinite

14. Self explanatory

15. Here, $v^2 = \omega^2(r^2 - y^2)$

$\therefore 9 = \omega^2(r^2 - 16)$,

And $16 = \omega^2(r^2 - 9)$

Dividing, we get $\frac{9}{16} = \frac{r^2 - 16}{r^2 - 9}$

Or $7r^2 = 175$

Or $r = 5m$

16. Given: $l_1 = 0.5m, l_2 = 2.0m$

Time period of simple pendulum, $T = 2\pi\sqrt{\frac{l}{g}}$

For $l_1 = 0.5m, T_1 = 2\pi\sqrt{\frac{0.5}{g}}$

And for $l_2 = 2.0m, T_2 = 2\pi\sqrt{\frac{2.0}{g}}$

$\Rightarrow \frac{T_2}{T_1} = \sqrt{\frac{2.0}{0.5}} = 2$

$\Rightarrow T_2 = 2T_1$

Hence both the pendulums will be in same phase when the shorter length pendulum completes 2 oscillations.

17. Phase difference = $\frac{2\pi}{\lambda} \times \text{path difference}$

$$\therefore \frac{\pi}{3} = \frac{2\pi}{\lambda} \times \text{path difference}$$

$$\Rightarrow \text{Path difference} = \frac{\lambda}{6}$$

18. Doppler effect is not observed when source and observer are moving in perpendicular direction.

19. Self explanatory

20. Efficiency of carnot engine, $\eta = 1 - \frac{T_2}{T_1}$

In case I $0.4 = 1 - \frac{T_2}{500}$

$$\Rightarrow \frac{T_2}{500} = 1 - 0.4$$

$$\Rightarrow T_2 = 500 \times 0.6 = 300^0K$$

In case II $0.5 = 1 - \frac{300}{T'_1}$

$$\Rightarrow \frac{300}{T'_1} = 1 - 0.5$$

$$\Rightarrow T'_1 = \frac{300}{0.5} = 600^0K$$

21. Mass and volume of the gas will remain same, so density will also remain same.

22. According to Newton's law of cooling if two liquids are cooled under identical conditions, then their rates of cooling will be same.

23. According to Wien's displacement law

$$\lambda_m = \text{constant}$$

$$\Rightarrow \lambda_m \propto \frac{1}{T}$$

$$\therefore \frac{(\lambda_m)'}{\lambda_m} = \frac{T_1}{T_2} = \frac{2000}{3000}$$

$$\text{Or } (\lambda_m)' = \frac{2}{3} \lambda_m$$

24. Resultant force = $\sqrt{F^2 + F^2 + 2F^2 \cos 60^0}$

$$= \sqrt{3F^2} = \sqrt{3}F$$

25. $V = 5x^2 + 10x - 9$

Differentiating, we get $\frac{dV}{dx} = 10x + 10$

At point $x = 1$; $\frac{dV}{dx} = 20$

But $\frac{dV}{dx} = E$, therefore $E = 20Vm^{-1}$

26. There will be two dipoles inclined to each other at an angle of 60^0 . The dipole moment of each dipole will be (ql)

\therefore Resultant dipole moment

$$= \sqrt{(ql)^2 + (ql)^2 + 2(ql)(ql) \cos 60^0} = \sqrt{3}(ql)$$

27. $R = \rho \frac{l}{\pi r^2}$ and $m = \pi r^2 ld$

$$\therefore R \propto \frac{l^2}{m}$$

$$\therefore R_1 : R_2 : R_3 = \frac{25}{1} : \frac{9}{3} : \frac{1}{5} = 125 : 15 : 1$$

28. $V_d = \frac{I}{neA}$

$$= \frac{1}{8 \times 10^{22} \times 1.6 \times 10^{-19} \times 7.81 \times 10^6}$$

$$= 10m/s$$

$$\therefore \frac{l}{V_d} = \frac{10 \times 10^{-1}}{10} = 10^{-2} \text{sec}$$

29. Self explanatory

30. Drift velocity, $v_d = \frac{I}{neA} = \frac{V}{neAR}$

Or $v_d \propto V$

So, when the potential difference is doubled, the drift velocity will be doubled.

31. $F = Bil \sin \theta$
 $= 2 \times 0.5 \times 5 \times 10^{-4} \times \sin 60^\circ$
 $= 4.33 \times 10^{-4} \text{N}$
32. Here $\frac{mv^2}{r} = Bqv$
 Or $v = \frac{Bqr}{m}$
 Or $r\omega = \frac{Bqr}{m}$
 Or $\omega = \frac{Bq}{m}$
 Or $\frac{2\pi}{T} = \frac{Bq}{m}$
 Or $T = \frac{2\pi m}{Bq}$
 But $T \propto \frac{m}{q}$
 $\therefore \frac{T_\alpha}{T_p} = \frac{m_\alpha}{M_p} \times \frac{q_p}{q_\alpha} = 4 \frac{m_p}{m_p} \times \frac{q_p}{2q_p} = 2$
33. Self explanatory
34. Charge in capacitor, $q = CV$
 $= 10^{-6} \times 200\sqrt{2} \sin 100t$
 or $i = \frac{dq}{dt} = 10^{-6} \times 200\sqrt{2} \times 100 \cos 100t$
 $t = 20\sqrt{2} \times 10^{-3} \cos 100t$
 \therefore Ammeter reading in rms value
 $= \frac{I_m}{\sqrt{2}} = 20\sqrt{2} \times \frac{10^{-3}}{\sqrt{2}} = 20 \text{mA}$
35. $X_c = \frac{1}{\omega C}$
 or $f = \frac{1}{2\pi C X_c}$
 $= 1 \times \frac{1000}{6.28 \times 5 \times 10^{-6}}$
 $= \frac{1000}{\pi} \text{MHz}$
36. Charge, $q = \omega CV = \omega CV_0 \sin \omega t$
 $\therefore i = \frac{dq}{dt} = \omega CV_0 \cos \omega t = \omega CV_0 \sin \left(\omega t + \frac{\pi}{2} \right)$
37. Phase difference, $\phi = \tan^{-1} \frac{\omega L}{R}$
 $= \tan^{-1} \frac{2\pi \times 50 \times 1.4}{440} = \tan^{-1} 1 = \frac{\pi}{4}$
 $\therefore \omega t = \frac{\pi}{4}$
 or $t = \frac{\pi}{4} \times \frac{1}{\omega} = \frac{\pi}{4} \times \frac{1}{2\pi \times 50} = 2.5 \times 10^{-3} \text{sec}$
38. Force due to current carrying wire
 $F = i(dl \times B) = ilB \sin \theta$
 Or $\sin \theta = \frac{F}{ilB} = \frac{15}{1.5 \times 10 \times 2} = \frac{1}{2}$
 Or $\theta = \sin^{-1} \left(\frac{1}{2} \right) = 30^\circ$
39. Distance between two consecutive dark fringes
 $= \frac{\lambda D}{d} = \frac{6000 \times 10^{-10} \times 1}{0.6 \times 10^{-3}}$
 $= 1 \times 10^{-3} \text{m} = 1 \text{mm}$
40. $\frac{I_1}{I_2} = \frac{100}{1}$
 Now, $\frac{I_{max}}{I_{min}} = \left[\frac{\sqrt{\frac{I_1}{I_2} + 1}}{\sqrt{\frac{I_1}{I_2} - 1}} \right]^2 = \left[\frac{\sqrt{100+1}}{\sqrt{100-1}} \right]^2 = \frac{121}{81} = \frac{3}{2}$
41. Using relation, $d \sin \theta = n\lambda$

$$\Rightarrow \sin \theta = \frac{n\lambda}{d}$$

$$\text{For } n = 3, \sin \theta = \frac{3\lambda}{d} = \frac{3 \times 589 \times 10^{-19}}{0.589}$$

$$= 3 \times 10^{-6}$$

$$\text{or } \theta = \sin^{-1}(3 \times 10^{-6})$$

42. 44. Self explanatory

$$45. E = \frac{V}{d} = \frac{0.5}{5 \times 10^{-7}} = 10^6 \text{Vm}^{-1}$$

46. Self explanatory

$$47. \text{Current flow is possible and } i = \frac{V}{R} = \frac{3}{300} = 10^{-2} \text{A}$$

48. 53. Self explanatory

$$54. S = \frac{I_g G}{I - I_g}$$

$$= \frac{10 \times 10^{-3} \times 55}{1 - \frac{1}{100}} = \frac{55}{99}$$

$$= 0.555 \Omega$$

$$55. \frac{r_e}{r_p} = \frac{m_e v_e}{q_e B} \times \frac{q_p B}{m_p v_p}$$

$$\text{Given: } m_e v_e = m_p v_p$$

$$\text{And } q_e = q_p$$

$$\therefore \frac{r_e}{r_p} = \frac{1}{1}$$

$$56. \oint \vec{B} \cdot d\vec{l} = \mu_0 I$$

$$= 4\pi \times 10^{-7} \times \frac{1}{4\pi}$$

$$= 10^{-7} \text{Weber/meter}$$

$$57. F = Bqv \sin 30^\circ$$

$$= 0.2 \times 1.6 \times 10^{-19} \times 10^7 \times \frac{1}{2}$$

$$= 1.6 \times 10^{-13} \text{N}$$

$$58. \omega = \frac{2\pi}{T} = \frac{2\pi}{0.1} = 20\pi$$

$$v_{max} = r\omega = 0.2 \times 20\pi = 4\pi \text{ms}^{-1}$$

$$59. l = \frac{\lambda}{4}, \Rightarrow \lambda = 4l$$

$$\therefore \text{Frequency, } n = \frac{v}{\lambda}$$

$$\text{Or } 264 = \frac{330}{4l}$$

$$\Rightarrow l = \frac{330}{264 \times 4} = \frac{330}{1056}$$

$$= 0.3125 \text{ meter} = 31.25 \text{ cm}$$

60. From first law of thermodynamics

$$dQ = dU + dw$$

$$\therefore 110 = dU + 40$$

$$\text{Or } dU = 110 - 40 = 70 \text{J}$$

CHEMISTRY

ANSWER KEYS

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4	3	1	1	2	1	3	3	3	3	1	1	1
53	54	55	56	57	58	59	60					
1	3	2	1	2	4	4	3					

HINTS & SOLUTIONS

- Zn rod is placed in 20% NH_4Cl solution
- 1 At.wt of bivalent metal required $2F = 2N$ electrons
- Threshold energy = Activation energy + Internal energy
- Increase in rate of reaction is maximum for the reaction having the maximum activation energy
- order = $\frac{3}{2} - 1 = \frac{1}{2}$
- $\log \frac{k_2}{k_1} = \frac{E_a}{2.303R} \left[\frac{1}{T_1} - \frac{1}{T_2} \right]$
- $\text{Cr}_2\text{O}_7^{2-} \rightarrow 2\text{Cr}^{+3} + 6e^-$ $E_{\text{naHCO}_3} = \frac{84}{1}$
- Particle concentration is same
- $\frac{P^0 - P}{P_0}$ = mole fraction of solute
- $\Delta T_b = K_b \times \frac{\text{Wt. of glucose}}{\text{molecular weight of glucose}} \times \frac{1000}{\text{Wt. of solvent}}$
- Boiling point depends on number of solute particles hence highest in 0.1 M FeCl_3
- $\Delta T_f = K_f \cdot m$
 $m = \frac{\Delta T_f}{K_f} = \frac{0.0054}{1.8} = 0.003$
 \therefore 0.001 molal compound produces 0.003 molal ions
 $\therefore [\text{Pt}(\text{NH}_3)_4\text{Cl}_2]\text{Cl}_2 \rightarrow [\text{Pt}(\text{NH}_3)_4\text{Cl}_2]^{2+} + 2\text{Cl}^-$
- Adsorption theory does not explain the acid-base catalysis
- Ore has lower density in froth floatation process
- Zone refining
- $\text{K}_2\text{Cr}_2\text{O}_7 + 4\text{NaCl} + 6\text{H}_2\text{SO}_4 \rightarrow 2\text{KHSO}_4 + 4\text{NaHSO}_4 + 2\text{CrO}_2\text{Cl}_2 + 3\text{H}_2\text{O}$ Chromyl chloride
- This corresponds to 3 unpaired electrons (Cr^{3+})
- Reactivity order of hydrogens is $3^\circ > 2^\circ > 1^\circ$
- $\text{C}_2\text{H}_5 - \text{Cl} \xrightarrow{\text{KCN}} \text{C}_2\text{H}_5 - \text{CH} \xrightarrow{\text{H}_3\text{O}^+}$
(A) Acid
- Pyrene (CCl_4) is used as fire extinguisher
- CHCl_3 is an anaesthetic agent
- Ethyl alcohol on dehydration at 140°C with con H_2SO_4 acids to give ethers

38. $\text{CH}_3\text{COONa} + \text{NaOH} \xrightarrow[\text{(Sodalime)}]{\text{CaO}} \text{CH}_4 + \text{Na}_2\text{CO}_3$
39. The abnormal trend of 3^o amines is explained in terms of steric effect. Note basic order of amines on the basis of pK_b, reported in Finar
- | | | | | | | | |
|-----------------|------------------------------------|---|---------------------------------|---|-----------------------------------|---|-----------------|
| | (CH ₃) ₂ NH | > | CH ₃ NH ₂ | > | (CH ₃) ₃ N | > | NH ₃ |
| pK _b | 3.23 | | 3.32 | | 4.2 | | 4.73 |
- | | | | | | | | |
|-----------------|--|---|---|---|---|---|-----------------|
| | (C ₂ H ₅) ₂ NH | > | (C ₂ H ₅) ₃ N | > | C ₂ H ₅ NH ₂ | > | NH ₃ |
| pK _b | 3.07 | | 3.13 | | 3.37 | | 4.73 |
40. Cellulose is most abundant in nature
42. $4.48 \text{ litre CH}_4 = \frac{4.48}{22.4} = \frac{1}{5} = 0.2 \text{ mol}$
 $= 16 \times 0.2 = 3.2 \text{ g CH}_4$
 $= 0.2 \times 6.02 \times 10^{23}$
 $= 1.2 \times 10^{22} \text{ molecules}$
43. Outermost electron of Rb (At. no. 37) is 5s. Hence, its quantum numbers are 5, 0, 0 ± ½.
44. This is as per statement of ionization energy
45. K⁺ (Z = 19), Ca²⁺ (Z = 20), Sc³⁺ (Z = 21) and Cl⁻ (Z = 17), all have 18 electrons each and as such are isoelectronic.
47. ${}^{\circ}CO_2 = \left(\frac{3RT_1}{44}\right)^{1/2}$
 ${}^{\circ}N_2 = \left(\frac{3R \times 294}{28}\right)^{1/2}$
 $\left(\frac{3RT_1}{44}\right)^{1/2} = \left(\frac{3R \times 294}{28}\right)^{1/2}$
 or $T_1 = 462 \text{ K or } 189^{\circ}\text{C}$
48. $T = 47 + 273 = 320 \text{ K}$
 K.E. = $\frac{3}{2} RT$ (For 1 mole)
 $= \frac{3}{2} \times (8.314 \text{ JK}^{-1} \text{ mol}^{-1}) \times (320 \text{ K})$
 $= 3990.7 \text{ J mol}^{-1}$
 Molar mass of O₂ = 32 g mol⁻¹
 $\therefore \text{K.E. (per gram of O}_2) = \frac{3990.7}{32} \text{ Jg}^{-1}$
 $= 124.7 \text{ Jg}^{-1}$
 $= 1.247 \times 10^2 \text{ Jg}^{-1}$
49. $\Delta n_g = 4 - 2 = +2$
 Thus $\Delta H > \Delta E$
50. As the equation is multiplied with 2, the equilibrium constant becomes square of the original value
51. Precipitation occurs only when ionic product exceeds solubility product.
 In case (A) ionic product
 $= 10^{-4} \times 10^{-4}$
 $= 10^{-8} > 1.8 \times 10^{-10}$
53. $\text{C}_3\text{C}_7\text{COONa} + \text{NaOH} \xrightarrow{\text{CaO}} \text{C}_3\text{H}_8 + \text{Na}_2\text{CO}_3$
55. Most hazardous metal pollutant of automobile exhaust is lead
56. Ca²⁺, K⁺ and S²⁻ are isoelectronic species and for these greater the nuclear charge (atomic number), smaller will be the size.
57. Tranquilizers affect the CNS system and induce sleep

58. As atomic number increases Halogen gains electrons less readily

59. Catalytic poison for Fe is H_2S .

60. $\frac{m_1}{m_2} - \frac{E_1}{E_2} = 250 \frac{m_1}{m_2} = \frac{E_1}{E_2}$

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UNIVERSAL ACADEMY

MATHEMATICS

ANSWER KEYS

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HINTS & SOLUTIONS

$$1. \quad y = \frac{2x+1}{3}$$

$$\frac{3y-1}{2} = x$$

$$\frac{3y-1}{2} = f^{-1}(y)$$

$$\frac{3x-1}{2} = f^{-1}(x)$$

$$2. \quad \frac{2^{x+3}}{2^{x-1}} = 2^4 = f(4)$$

$$3. \quad f(x+y) = f(x)f(y)$$

$$\Rightarrow f(x) = P^x$$

$$f(5) = 32 = P^5$$

$$\Rightarrow P = 2$$

$$f(7) = 2^7 = 128$$

$$4. \quad 25 - x^2 \geq 0$$

$$x^2 \leq 25$$

$$5. \quad \sin \cot^{-1} \left(\frac{1}{\sqrt{x^2+1}} \right)$$

$$= \frac{\sqrt{x^2+1}}{\sqrt{x^2+2}}$$

$$6. \quad \frac{\pi}{4} = \tan^{-1} 1$$

$$2 \tan^{-1} x = \tan^{-1} \left(\frac{2x}{1-x^2} \right)$$

7. Standard result

8. Standard result

9. $\text{Tr}(A-2B)$

$$= \text{Tr}(A) - 2 \text{Tr}(B)$$

10. Standard result

11. $|\text{Adj } A| = |A|^{n-1}$

13. $C_1 \rightarrow C_1 - C_2$

$$(a + b)^2 + (a - b)^2 = 4ab$$

14. Determinant of skew-symmetric

Matrix of order 3 is 0

15. $\lim_{x \rightarrow 1} \frac{\log x}{x-1} = K$

$$\frac{1}{1}$$

$$\lim_{x \rightarrow 1} \frac{x}{1} = K \quad (\text{L.H. Rule})$$

$$1 = K$$

16. $\lim_{x \rightarrow 1} x \sin\left(\frac{1}{x}\right) = 0 = f(0)$

\therefore It is continuous

$$f(0) = \lim_{h \rightarrow 0} \frac{f(0+h) - f(0)}{h} = \lim_{h \rightarrow 0} \sin\left(\frac{1}{h}\right) = \text{does not exist}$$

17. $h(x) = e^{\sin^{-1} x}$

$$h'(x) = e^{\sin^{-1} x} \times \frac{1}{\sqrt{1-x^2}}$$

18. $x = \frac{\sin y}{\sin(a+y)}$

$$\frac{dy}{dx} = \frac{\sin(a+y-y)}{\sin^2(a+y)}$$

$$\frac{dy}{dx} = \frac{\sin^2(a+y)}{\sin a}$$

19. Take $x = \cos \theta$

$$y = \frac{1}{2}(1 - \cos \theta)$$

$$y = \frac{1}{2}(1 - x)$$

20. $g(x) = f^{-1}(x)$

$$\Rightarrow f[f(x)] = x$$

$$\Rightarrow g'(f(x)) \cdot f'(x) = 1$$

$$f(0) = 1 \quad f'(0) = \frac{1}{2}$$

$$g'(f(0)) = \frac{1}{f'(0)}$$

$$g'(1) = 2$$

21. $\frac{dy}{dt} = 2 \frac{dx}{dt}$

$$y^2 = 18x$$

$$2y \frac{dy}{dt} = 18 \frac{dx}{dt}$$

$$\Rightarrow y = \frac{9}{2} \text{ and } x = \frac{9}{8}$$

$$22. y = y_1 = -\frac{1}{m}(x - x_1)$$

$$m = \frac{dy}{dx}(a, a)$$

$$23. f(x) = 1 - \sin^2 x \cos^2 x$$

$$f(x) = 1 - \frac{1}{2} \sin^2 2x$$

$$f'(x) = 0 \Rightarrow x = \frac{\pi}{4}, \frac{\pi}{2}$$

$$f''\left(\frac{\pi}{4}\right) > 0$$

$$f''\left(\frac{\pi}{4}\right) = \frac{1}{2}$$

$$24. f'(x) = \frac{-(2+x)}{(1+x)^2} < 0, \text{ for } x > 0$$

$$25. \text{ By parts } \frac{d}{dx} \left(\log \tan \left(\frac{1}{2} \right) \right) = \frac{1}{\sin x}$$

$$26. \int x^{n-1} \left(\frac{1}{x^n} - \frac{1}{x^n + 1} \right) dx$$

$$\int \left(\frac{x^{n-1}}{x^n} - \frac{x^{n-1}}{x^n + 1} \right) dx$$

$$\frac{1}{n} [\log x^n - \log(x^n + 1)] + c$$

$$27. \cos x + \sqrt{3} \sin x$$

$$= 2 \left(\cos x \cos \frac{\pi}{3} + \sin x \sin \frac{\pi}{3} \right)$$

$$= 2 \cos \left(x - \frac{\pi}{3} \right)$$

$$\therefore I = \frac{1}{2} \int \sec \left(x - \frac{\pi}{3} \right) dx$$

$$28. \int \sec^2(xe^x) \cdot \frac{d}{dx}(xe^x) dx$$

$$29. \int_0^{\frac{\pi}{4}} \sin^n x = \int_0^{\frac{\pi}{4}} \cos^n x$$

$$30. \text{ Standard result}$$

$$31. \log_e^x = t \Rightarrow e^t = x$$

$$\Rightarrow e^t dt = dx$$

$$32. \text{Area} = \int_1^2 (x^2 + 2) dx$$

$$33. x^3 = \sqrt{x} \Rightarrow x^6 = x \\ \Rightarrow x = 0 \text{ or } 1$$

$$\text{Area} = \int_0^1 (\sqrt{x} - x^3) dx$$

$$34. \text{Area} = \frac{(a)(a)}{3} = 1$$

$$35. \left(\frac{d^2 y}{dx^2} \right)^3 = y^4$$

$$36. \int y dy = \int (c - x) dx + K \\ x^2 + y^2 - 2cx = 2K$$

37. Divided by $\cos x$

$$\Rightarrow \frac{dy}{dx} + \tan x dx = \sec x$$

$$I.F = \int \tan x dx = e^{\log \sec x} = \sec x$$

38. Common between $A \cup B$ and A is A

$$39. \tan \theta = \cot(90 - \theta) \text{ and } \tan \theta \times \cot \theta = 1$$

$$40. P(4) : 2^4 < 1 \times 2 \times 3 \times 4 \text{ is true}$$

41. Standard result

$$42. {}^{12}C_3 - {}^7C_3$$

43. Among 52 terms 26 terms are cancelled

44. A.M \geq G.M

$$\frac{3^x + 3^{1-x}}{2} \geq \sqrt{3^x \times 3^{1-x}}$$

$$45. x + y = 5$$

$$3 + 2 = 5$$

$$46. r = \sqrt{(4-1)^2 + (6-2)^2}$$

$$r = 5, A = \pi r^2$$

$$47. \lim_{x \rightarrow 0} \frac{\frac{1 - \cos x}{\sin x}}{x}$$

$$\lim_{x \rightarrow 0} \frac{1 - \cos x}{x \sin x}$$

$$\lim_{x \rightarrow 0} \frac{2 \sin^2 \frac{x}{2}}{x \times 2 \sin \frac{x}{2} \times \cos \frac{x}{2}}$$

$$\lim_{x \rightarrow 0} \frac{\sin\left(\frac{x}{2}\right)}{\frac{x}{2} \times 2} = \frac{1}{2}$$

48. Given $p \wedge q$

$$N(p \wedge q) = Np \vee Nq$$

49. Standard result

$$50. \frac{1}{2} |\vec{AB} + \vec{AC}|$$

$$51. a \cdot (\vec{b} \times \vec{c}) = (\vec{c} \times \vec{a}) \cdot \vec{b}$$

$$\vec{b} \cdot (\vec{c} \times \vec{a}) = (\vec{a} \times \vec{b}) \cdot \vec{c}$$

$$\vec{c} \cdot (\vec{a} \times \vec{b}) = (\vec{b} \times \vec{c}) \cdot \vec{a}$$

52. Substitute a point in options and verify

$$53. 1 = \frac{a+1+2}{4} \Rightarrow a = 1$$

$$2 = \frac{2+b+1}{4} \Rightarrow b = 5$$

$$-1 = \frac{3+2+c}{4} = 1c = -9$$

$$\sqrt{a^2 + b^2 + c^2} = \sqrt{81 + 25 + 1} = \sqrt{107}$$

$$54. \frac{h - x_1}{a} = \frac{-(ax_1 + by_1 + cz_1 + d)}{a_2 + b_2 + c_2} = \frac{k - y_1}{b} = \frac{l - z_1}{c}$$

$$(a, b, c) = (2, -1, 1)$$

$$(x_1, y_1, z_1) = (1, 3, 4)$$

Find h, k and l

55. Standard result

56. Standard result

$$57. P(B \cap A') = P(B) - P(A \cap B) \\ = P(A \cup B) - P(A)$$

$$P\left(\frac{B}{A'}\right) = \frac{P(B \cap A')}{P(A')}$$

$$58. P(\text{red or black}) = \frac{3+7}{3+4+7} \\ = \frac{10}{14} = \frac{5}{7}$$

$$59. P(A \cap B) = P(A)P(B)$$

$$P(A \cup B) = P(A) + P(B) - P(A \cap B)$$

$$P(A \cup B) = P(A) + P(B) - P(B)$$

$$60. \text{ Selection of one winning horse is } \frac{1}{5}$$